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**NATIONAL STANDARD OF THE PEOPLE'S REPUBLIC
OF CHINA**

中华人民共和国国家标准

GB/T 23513.1-2009

**Chemical analysis methods for germanium
concentrate—Part 1: Determination of
germanium content—Potassium iodate titration**

锗精矿化学分析方法

第 1 部分：锗量的测定 碘酸钾滴定法

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Foreword

GB/T 23513 Chemical analysis methods for germanium concentrate consists of five parts are as follows:

- Part 1: Determination of germanium content—Potassium iodate titration;
- Part 2: Determination of arsenic content—Ferrous ammonium sulfate titration;
- Part 3: Determination of sulfur content—Barium sulfate gravimetry;
- Part 4: Determination of fluoride content—ISE;
- Part 5: Determination of silica content—Gravimetry

This is part one.

This part is proposed by China Nonferrous Metals Industry Association.

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The responsible drafting organizations of this part are Yunnan Lincang Xinyuan Germanium Co., Ltd.

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Chemical analysis methods for germanium concentrate--

Part 1: Determination of germanium content—Potassium iodate titration

1 Scope

This part of GB/T 23513 specifies the determination method of germanium content in germanium concentrate.

This part applies to the determination of germanium content in germanium concentrate.

Determination range: 1.0%-70%.

2 Methods and principles

Melt the sample with sodium hydroxide and use phosphoric acid and potassium permanganate to prohibit the escape of distilled arsenic, distilled antimony, distilled tin and others. Take hypophosphite to return tetravalent germanium to divalent germanium and take starch as an indicator in 3 mol/L of phosphoric acid and 4.5 mol / L of hydrochloric acid and titrate it with standard potassium iodate solution.

3 Reagents

Only reagents after pure analysis and distilled water or water with equivalent purity can be used in the analysis unless otherwise stated

3.1 Potassium iodide (KI).

3.2 Anhydrous sodium carbonate (Na_2CO_3).

3.3 Sodium hydroxide (NaOH).

3.4 Hypophosphite ($\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$)

3.5 Potassium permanganate (KMnO_4).

3.6 Ammonium fluoride (NH_4F).

3.7 Hydrochloric acid (ρ 1.19 g / mL^{-1}).



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