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GB/T 14849.2-2007 Replace GB/T 14849.2-1993

Methods for chemical analysis of silicon metal-Part 2: Determination of aluminum content—Chrome azurol S spectrophotometric method

工业硅化学分析方法 第 2 部分: 铝含量的测定铬天青-S 分光光度法

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Foreword

GB/T 14849 Methods for Chemical Analysis of Silicon Metal is divided into four parts:

----Part 1: Determination of iron content - 1, 10-phenanthroline spectrophotometric method: ----Part 2: Determination of aluminum content - chrome azurol-S spectrophotometric method; ——Part 3: Determination of calcium content - flame atomic absorption spectrometry, chlorophosphonazo I spectrophotometric method; ——Part 4: Inductively coupled plasma-atomic emission spectrometry to determine element content. This is Part 2 of GB/T 14849. This Part will replace GB/T 14849.2-1993 Silicon metal--Determination of aluminium content--Chrome azurol S spectrophotometric method, comparison with GB/T 14849.2-1993, main changes are as follows: ——"repeatability" and "quality assurance and control" is added. This Part is proposed by China Nonferrous Metals Industry Association. This Part is under the jurisdiction of the National Technical Committee on Nonferrous Metals of Standardization. This Part is drafted by the unit of Fushun Aluminium Co., Ltd. Main drafters of this Part: Xu Tieling, Ji Chunlei, Yang Yuhong and Xu Tieling. The release conditions of previous versions of the standards replaced by this Part are as follows:

——GB/T 14849.2-1993

Methods for chemical analysis of silicon metal-- Part 2: Determination of aluminum content—Chrome azurol S spectrophotometric method

1 Scope

This Part specifies the method for determination of aluminum content in silicon metal.

This Part applies to the determination of aluminum content in silicon metal. Determination range (quality fraction): 0.02%~0.30%

2 Method summary

The specimen is decomposed with hydrofluoric acid and nitric acid, sulfuric acid smoking removes silicon, fluorine, etc. and the residue is dissolved with hydrochloric acid. Ascorbic acid is used to mask the interference of ferrum. A purplish red complex compound is formed by the reaction between aluminum and chrome azurol-S in hexamethylenetetramine medium of pH5.5-6.1. Use the spectrophotometer with wavelength of 545 nm to determine the absorbance.

3 Reagents

- **3.1** Hydrofluoric acid (ρ 1.14g/mL)
- 3.2 Nitric acid (1+1)
- **3.3** Sulfuric acid (1+1)
- **3.4** Hydrochloric acid (1+1)
- 3.5 Ascorbic acid solution (10g/L). Prepare it when use it
- **3.6** Hexamethylenetetramine solution (300g/L)
- **3.7** Chrome azurol S acetic acid solution (0.3g/L): 0.30g chrome azurol-S is put into a beaker, and 25mL water and 25mL anhydrous alcohol are added. 475mL water is added after dissolution, and the homogenized sample is diluted to 1000mL with anhydrous alcohol, evenly mixed.
- **3.8** Aluminum standard storage solution: 0.2500g metal aluminum is put into a polyethylene cup. Then about 20mL water and 3.0g sodium hydroxide is added. After the reaction slows down, heat it on the water bath until dissolving completely. Hydrochloric acid (3.4) is used for slow neutralization until precipitation appears. Then excessive 20mL



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